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International Lower Secondary

Student's Book

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SCIENCE

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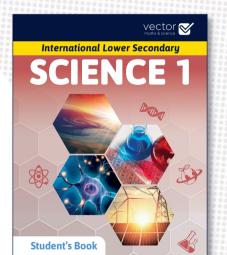
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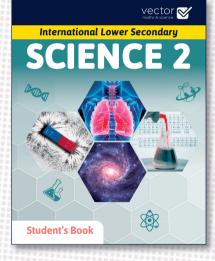
International Lower Secondary SCIENCE Levels 1-3

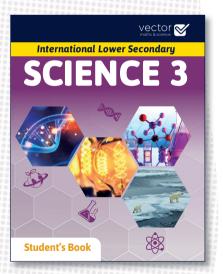
VECTOR M&S International Lower Secondary Science series is a complete three-level course aligned to international standards of education, designed to meet the needs of secondary school students. This series aims to promote a deep understanding of scientific concepts in a friendly environment. A number of practical and scientific skills are developed through lessons that stimulate the interest of students and make connections between science and real life.

This series will provide students with an in-depth understanding of scientific concepts and theories, preparing them for success in their future studies by achieving a set of learning objectives in the scientific domains of Biology, Chemistry, Physics as well as Earth and Space.



Biology





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Earth and Space

Physics

Chemistry

Key features

FOR STUDENTS:

 cover pages with information on real-life situations, learning objectives of the unit and questions on prior knowledge

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- high-quality illustrations and photographs
- a wide variety of discussion topics that promote interactive learning
- numerous activities to practise what was learnt in the theory sections, develop scientific enquiry skills through exploration and also challenge the learners with extra information on the topic taught
- review activities that allow the assessment of students' progress
- a section with the highlights at the end of each unit
- an extensive glossary to facilitate the development of students' vocabulary as well as a list with pictures of the laboratory apparatus for learners to become more familiar with the experiment procedure
- a section with information on methods learners should use as scientists
- a list of the keywords and an overview of the lesson at the end which help learners summarise what they have covered in the corresponding lesson

FOR TEACHERS:

- a detailed map of the Student's Book, Workbook and Teacher's Book that helps the teacher understand the structure of each book
- an extensive section with extra theory on the concepts taught in the unit and the corresponding keywords along with keyword information points for the teacher's reference
- step-by-step lesson plans for each unit
- digital resources such as Worksheets and Resource Sheets to support comprehension and extension of knowledge
- optional activities for further practice
- safety warnings and guidelines
- keys for all the questions and activities in the Student's Book, the Workbook and the resources as well as detailed steps for the implementation of the experiments

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Components

FOR STUDENTS



FOR TEACHERS

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Science 1

| Lessons | Learning objectives | | |
|---|---|--|--|
| Unit 1: Energy | Unit 1: Energy | | |
| Energy stores and transfers Energy transformations Energy is conserved Reducing wasted energy | Identify the ways in which energy can be stored. Describe how energy can be transferred. Describe energy transformations. State the principle of conservation of energy and apply it in everyday situations. Identify that during energy transformations some of the energy will be useful and the rest will be wasted. | | |
| | Compare the energy that is being wasted by different devices. | | |
| Unit 2: Properties of matter | | | |
| 2.1 The states of matter (particle model) 2.2 How matter changes states 2.3 The water cycle 2.4 Mixtures 2.5 Elements and compounds 2.6 Atoms and molecules 2.7 Chemical symbols and formulae | Identify and describe how particles are arranged and move in the three states of matter. Relate the different properties of matter to the particle model. Identify the lack of particles as a vacuum. Describe the changes of state using the particle model. Identify the stages of the water cycle and explain how water is recycled. Identify that air is a mixture of different gases. Classify mixtures as heterogeneous or homogeneous according to their properties, and explore ways to separate them. Describe what a compound is and how it differs from an element and from a mixture. Define atoms and molecules, and identify molecules of compounds and elements. Describe the Periodic Table as a way to sort elements. Identify and write the chemical symbols of the first twenty elements of the Periodic Table. Write the formulae of simple compounds, name the compounds and determine the elements and the number of atoms in compounds. | | |

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| Unit 3: Cells | |
|---|---|
| 3.1 All living things are ma of cells | de • Describe how an organism is made up of organs, organs are made from tissues and tissues from cells. |
| 3.2 Observing cells | Describe how a group of organs works together to form an organ system. |
| 3.3 Plant and animal cells | Recognise that all living organisms are made of cells. |
| 3.4 Specialised cells | Use a microscope to observe cells. |
| | Name the basic parts of a cell and describe their functions. |
| | Name the parts that only plant cells contain and describe their functions. |
| | Distinguish between a plant cell and an animal cell. |
| | Name some specialised cells and describe how these specialised cells are adapted to |
| | perform their functions. |
| Unit 4: Solar system | |
| 1.1 Gravity | • Define the force of gravity as the force of attraction between any two objects. |
| 4.2 Our solar system | • Describe how the force of gravity depends on the masses of interacting objects and the |
| 4.3 Formation of the solar | distance between them. |
| system | Distinguish between mass and weight. |
| 4.4 The Sun, the Earth and | Recognise that gravity is the force that causes planets to orbit stars and moons to orbit |
| the Moon 4.5 Tides | planets. |
| t.J Hues | Identify that no air resistance acts on celestial bodies to affect their movement. |
| | Describe how the solar system formed from a nebula. |
| | Identify the phases of the Moon and explain the positioning of the Earth, the Moon and the Sun in each phase. |
| | |
| | Describe how solar and lunar eclipses occur and model them. |
| | Describe how solar and lunar eclipses occur and model them. Identify the causes of tides and explain why sea levels change between high and low tide. |
| Unit 5: Metals and non-me | • Identify the causes of tides and explain why sea levels change between high and low tide. |
| | Identify the causes of tides and explain why sea levels change between high and low tide. tals |
| 5.1 Metals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. |
| 5.1 Metals 5.2 Non-metals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. |
| 5.1 Metals 5.2 Non-metals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. |
| 5.1 Metals 5.2 Non-metals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal |
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| 5.1 Metals 5.2 Non-metals 5.3 Alloys | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Jnit 6: Acids and alkalis | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made or Explain why metals and their alloys have different properties using the particle theory. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 5.1 Acids and bases | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made of Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made of Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made of Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. |
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| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made of Explain why metals and their alloys have different properties using the particle theory. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life Unit 7: Classification of live | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made or Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. Describe how the strength of acids and bases can be measured using the pH scale. Describe everyday situations where acids and bases are involved. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life Unit 7: Classification of live 7.1 The seven life processe | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made or Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. Describe how the strength of acids and bases can be measured using the pH scale. Describe everyday situations where acids and bases are involved. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life Unit 7: Classification of live 7.1 The seven life processes 7.2 Classifying living thing 7.3 Classifying animals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made or Explain why metals and their alloys have different properties using the particle theory. • Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. Describe how the strength of acids and bases can be measured using the pH scale. Describe everyday situations where acids and bases are involved. |
| 5.1 Metals 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life Unit 7: Classification of live 7.1 The seven life processes 7.2 Classifying living thing 7.3 Classifying animals 7.4 Species | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made or Explain why metals and their alloys have different properties using the particle theory. Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. Describe how the strength of acids and bases can be measured using the pH scale. Describe everyday situations where acids and bases are involved. Ing things Identify living, once alive and never alive things based on the seven life processes. Classify organisms into plants and animals and these into smaller groups. Classify organisms into vertebrates and invertebrates and these into smaller groups. |
| 5.2 Non-metals 5.3 Alloys Unit 6: Acids and alkalis 6.1 Acids and bases 6.2 Detecting acids and alkalis 6.3 How acids and alkalis affect our everyday life Unit 7: Classification of live 7.1 The seven life processe 7.2 Classifying living thing 7.3 Classifying animals | Identify the causes of tides and explain why sea levels change between high and low tide. tals List the physical properties of metals. List the physical properties of non-metals. Compare metals and non-metals based on their physical properties. Define alloys as mixtures of two or more different elements, at least one of which is a metal Compare the properties of different alloys to each other and to the metals they are made of Explain why metals and their alloys have different properties using the particle theory. • Name some properties of acids and bases. Use indicators to identify whether a solution is acidic, alkaline or neutral. Describe how the strength of acids and bases can be measured using the pH scale. Describe everyday situations where acids and bases are involved. |

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| Unit 8: Sound waves | |
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| 8.1 How sound travels8.2 Reflection of sound8.3 Speed of sound | Describe how sound is transmitted by waves, and explain why sound does not travel through a vacuum. Describe how sound reflects off a surface creating an echo. Begin to realise the meaning and usefulness of the law of reflection. Identify and explain how the medium affects the speed of sound using the particle model. Calculate the speed of sound using reflected sound waves (echoes). |
| Unit 9: The Earth | |
| 9.1 Types of rocks9.2 Dating rocks9.3 Earth structure9.4 How Earth changes | Describe how igneous, sedimentary and metamorphic rocks are formed. Describe how fossils form. Explain how fossils are used to compare the age of rocks but cannot be used to show the age of the Earth. Describe the model of the structure of the Earth. Identify that the Earth's surface changes over time due to the movement of tectonic plates Describe how the movement of tectonic plates creates three types of plate boundaries. Identify that the movement of tectonic plates causes earthquakes, volcanoes and the formation of fold mountains at the boundaries of the plates. |
| Unit 10: The role of microorga | nisms in feeding relationships |
| 10.1 Feeding relationships 10.2 Microorganisms and viruses 10.3 Microorganisms affect food 10.4 Decomposers and their role | Use food chains and food webs to describe feeding relationships and the flow of energy in an ecosystem. Describe different types of microorganisms. Explain how viruses are different from microorganisms. Identify that microorganisms have an important role in decay. Investigate how temperature and moisture affect decomposition. Describe why decomposers are important to the environment. |
| Unit 11: Electricity | |
| 11.1 Electrical circuits 11.2 Electric current 11.3 Measuring the size of an electric current 11.4 Changing the number of cells in a circuit 11.5 Changing the number of bulbs in a series circuit 11.6 Electrical conductors and insulators | Relate the components of a circuit to their symbols. Construct circuit diagrams from circuits and vice versa. Use a model to show that an electric current is electrons flowing in one direction around a circuit. Measure electric current in a series circuit with an ammeter. Investigate how adding cells in series to a circuit affects the electric current. Investigate how adding bulbs in series to a circuit affects the electric current. Describe what electrical conductors and insulators are. Test materials to classify them as electrical conductors or insulators. Give examples of uses of electrical conductors and insulators in everyday life. |

| 2.1 | Introduction to chemical | Distinguish between physical and chemical changes. |
|-----|--|--|
| | reactions | Understand that new substances are formed by chemical reactions. |
| 2.2 | Ways to represent a chemical reaction | • Recognise that changes in temperature and colour are indicators of chemical reactions taking place. |
| 2.3 | Neutralisation reactions | Use word equations to describe chemical reactions. |
| 2.4 | Reactions that produce gases | Use the particle model to show how atoms rearrange themselves during a chemical reaction |
| 2.5 | Precipitation reactions | Describe a neutralisation reaction using a general word equation. |
| | | Describe how the pH changes during the neutralisation reaction. |
| | | Give examples of uses of neutralisation in everyday life. |
| | | • Recognise that the formation of a gas is an indicator of a chemical reaction taking place |
| | | Use different methods to identify the gases hydrogen, oxygen and carbon dioxide when given off in reactions. |
| | | Describe a precipitation reaction as a chemical reaction between solutions that forms a insoluble product. |
| | | Recognise that the formation of a precipitate is an indicator of a chemical reaction takin place. |

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Science 2

| | Lessons | Learning objectives |
|---------------------------------|---|--|
| Uni | it 1: Forces and motion | |
| 1.1 1.2 1.3 1.4 1.5 | Variables and measurements in experiments Speed Moving at constant speed The effects of forces on motion The turning effect of forces | Use appropriate equipment for the task to measure length and time and take those measurements with precision. Repeat measurements, calculate their mean and identify that this is more accurate than using just one result. Define speed as a measure of the distance moved by an object in a unit of time and identif metres per second (m/s) and kilometres per hour (km/h) as units for speed. Use the speed equation (speed = distance/time) to calculate one of the quantities, in the correct units, given the other two. Draw distance-time graphs for objects that move at a constant speed, and use them to calculate speed from the gradient and to compare the speed of objects. Analyse a distance-time graph for an object that changes speed. Use force diagrams to identify whether forces are balanced or unbalanced and describe how balanced or unbalanced forces affect the movement of objects. Recognise that the moment of force is the turning effect of a force about a specific point, calculate the moment of a force about a point and identify its units. Describe how an object is balanced when the moments acting clockwise are equal to the moments acting anti-clockwise. |

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Unit 2: Mixtures and solutions

| Mixtures | Distinguish between homogeneous and heterogeneous mixtures. |
|---------------------|---|
| Solutions | Define the terms solvent, solute and solution. |
| Solubility | Explore how mass is conserved during dissolving. |
| Factors that affect | • Explain the difference between dissolving and melting. |
| solubility | • Explain what happens during dissolving using the particle model. |
| 5 Chromatography | Give examples of solvents other than water. |
| | • Explain how the concentration of a solution is determined. |
| | • Define the solubility of a solvent and explain when the solution becomes saturated. |
| | Compare the solubility of different solutes in water. |
| | List factors that affect solubility. |
| | Explore how temperature affects solubility. |
| | Describe the process of paper chromatography. |
| | Analyse simple chromatograms. |

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| 3.3 3.4 | Diffusion Pressure on a surface Pressure in liquids Pressure in gases Upthrust - Sink or float | Use the particle theory to explain how diffusion happens in liquids and gases. Explore how temperature affects the rate of diffusion. Calculate pressure by dividing the force exerted on an area by the size of the area. Identify newtons per metre squared (N/m²) as a unit of measuring pressure. Relate pressure to force and area to describe everyday examples of high and low pressure. Describe what causes pressure in a liquid. Explore how depth affects the pressure in a liquid. Explain how hydraulic systems transmit forces. Describe what causes pressure in gases. Explain how the temperature and the concentration of a gas affect the pressure. |
|------------|--|---|
| | | Explain now the temperature and the concentration of a gas ancet the pressure. |

• Describe how atmospheric pressure varies with altitude.

• Explain how liquid pressure causes upthrust.

Unit 4: The respiratory system

| 4.1 | The human respiratory system | Identify the parts of the respiratory system and describe their functions. Describe how the diaphragm and the ribs move when you breathe in and out. |
|-----|---|--|
| 4.3 | Inhaling and exhaling Gas exchange in the lungs | Describe the structure of the alveoli and outline the process of gas exchange in them. Name the main components of blood as being red blood cells, white blood cells, plasma, and platelets, and describe their functions. |
| 4.5 | Blood Respiration in cells Health and smoking | Explain how red and white blood cells are adapted to perform their functions. State that aerobic respiration occurs in the mitochondria of the cells and energy is released. Describe the process of aerobic respiration using a word equation. Distinguish between respiration and breathing. Highlight the harms of smoking on human health. |

Unit 5: Atoms

| 5.1 | Developing models of the atom | • Explain how the atomic model has changed over time by describing different models of the atom. |
|-----|-------------------------------|---|
| 5.2 | The structure of the atom | Name the subatomic particles that make up the atom. |
| 5.3 | Pure substances | Describe the structure of the atom. |
| | | Describe the subatomic particles that make up the atom. |
| | | • Identify that the electrostatic attraction between the positive charges of protons and the negative charges of electrons holds the atom together. |
| | | Explain why atoms are electrically neutral. |
| | | • Explain what a pure substance is, and state the difference between its everyday meaning and its meaning in chemistry. |
| | | • Describe what an impurity is and give examples of common impurities found within substances. |
| | | Calculate the percentage purity of a substance. |
| | | • Describe how the purity of a substance can be determined by its physical properties. |
| | | Identify methods of separating and obtaining a pure substance. |
| | | • Describe how the purity of a substance can be determined by its physical properties. |

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| 6.1 The layers of the atmosphere | Describe how the Earth's atmosphere has changed since the formation of the Earth untinow. |
|----------------------------------|--|
| 6.2 Weather | Name the layers of the atmosphere in order. |
| 6.3 Climate | Identify the main gases that make up the atmosphere. |
| 6.4 Climate change | |
| 6.5 Global warming | Identify different types of weather. |
| | Use different equipment to record and describe the weather. |
| | Distinguish between weather and climate. |
| | • Describe how the climate in an area is affected by its latitude, altitude, and distance from the sea. |
| | Name the main climate zones on Earth and describe their climate conditions. |
| | • Evaluate data and evidence for climate change in the past. |
| | Describe some natural factors that cause climate change. |
| | Describe the greenhouse effect and the human activities that enhance it. |
| | Indicate the effects of climate change. |
| | List ways to reduce climate change. |
| | |
| | Identify the advantages and disadvantages of using different renewable and non-renewa resources. |
| | |
| Unit 7: Ecosystems and habita | its |
| 7.1 Ecosystem | • Define what an ecosystem is, and name some of the ecosystems on Earth. |
| 7.2 Different habitats | Define biotic and abiotic factors. |
| 7.3 Changes to ecosystems | Describe the interactions between biotic factors and those between biotic and abiotic |
| | factors of an ecosystem. |
| | • Explain the difference between a habitat and an ecosystem. |
| | Identify different habitats in an ecosystem. |
| | Describe how living things are adapted to their habitat. |
| | |
| | Use sampling to collect information about the organisms that live in a habitat. |
| | Identify changes in an ecosystem and their causes. |
| | • Describe the effect of toxic substances, such as DDT, being present in an ecosystem. |
| | Describe how invasive species have a negative effect on an ecosystem. |
| Unit 8: Light | |
| 8.1 Light travels in a straight | Demonstrate that light travels in straight lines using ray diagrams. |
| line | Distinguish between transparent and opaque materials. |
| 8.2 Reflection of light | • Explain why shadows are formed. |
| 8.3 Refraction of light | Describe how light is reflected off a flat surface. |
| 8.4 Dispersion of light | State the law of reflection. |
| 8.5 Colours | Draw and interpret ray diagrams of reflections of the light off flat surfaces. |
| | |
| | Distinguish between regular reflection off flat surfaces and diffuse reflection off irregula surfaces. |
| | • Explore how light is refracted when it passes from one transparent medium to another. |
| | • Explain that refraction happens because light travels at different speeds in different med |
| | Draw and interpret ray diagrams of refraction of light when travelling from one medium another. |
| | • Explore how white light is separated into different colours using a prism and name the |
| | seven colours of the visible spectrum in order. |
| | Classify coloured light into primary and secondary colours. |
| | Explain how filters work by subtracting light. |
| | Describe how the selective absorption and/or reflection of light affects the observed |
| | colour of an object. |

| 9.1 Asteroids | Identify what asteroids are and describe how they are formed. |
|---|--|
| 0.2 Galaxies | • Explore how the size of the asteroid and its speed affect the size of the crater it forms. |
| 0.3 Exploring the universe | Describe what a galaxy contains. |
| | • Identify that our solar system is part of a spiral galaxy called the Milky Way. |
| | • Classify galaxies as spiral, elliptical or irregular. |
| | • Identify that a light year is a unit used to measure astronomical distances and define the |
| | light year as the distance travelled by a beam of light in one year. |
| | Calculate the length of a light year in kilometres. |
| | • Describe how the universe began and how it changes over time. |
| Unit 10: Reactions | |
| 0.1 Reactions between | Describe the reactions between metals and acids, and recognise that some metals do not |
| metals and acids 0.2 Reactions between | react with acids. |
| metals and oxygen | Compare the reactivity of some metals with acids. |
| 0.3 Reactions between | Describe the reactions between metals and oxygen. Compare the reactivity of some metals with ovygen |
| metals and water | Compare the reactivity of some metals with oxygen. Describe the reactions between metals and water, and identify some inert metals. |
| 0.4 Endothermic reactions | Compare the reactivity of some metals with water. |
| and processes 0.5 Exothermic reactions and | Define what endothermic reactions are. |
| processes | Distinguish between endothermic reactions and processes. |
| processes | Give examples of how endothermic reactions and processes are used in everyday life. |
| | Define what exothermic reactions are. |
| | Destinguish between exothermic reactions and processes. |
| | Give examples of how exothermic reactions are used in everyday life. |
| | • Determine whether a reaction is exothermic or endothermic by measuring the temperature change. |
| Jnit 11: Nutrition and health | |
| 1.1 Human growth and | Describe how fertilisation creates a zygote that develops into an embryo. |
| development 1.2 Nutrients | • Distinguish between growth and development and describe their stages in the human life cycle. |
| 1.3 Energy needs | • Define nutrients and highlight the importance of water and fibre for our bodies. |
| 1.4 Effects of a poor diet | • Identify carbohydrates, proteins, fats, vitamins, and minerals as nutrients, give examples |
| 1.5 How the body moves | of foods that contain them and explain how they help our bodies. |
| | Use food labels to identify the ingredients and energy in foods. |
| | • Explain that having a balanced diet means eating food containing all nutrients in the righ amounts. |
| | • Explain that organisms require energy for essential processes in an amount depending on their activity, and state that food is a store of energy measured in kilojoules (kJ) or kilocalories (kcal). |
| | • Describe health problems and deficiency diseases caused by having an unbalanced diet, and highlight the importance of exercise for human health. |
| | Describe how antagonistic muscle pairs work together. |
| | • Describe the movement in hinge, and ball-and-socket joints. |

Unit 12: Magnetism

- 12.1 Properties of magnets
- 12.2 Magnetic fields
- 12.3 Electromagnets
- 12.4 Uses of electromagnets
- Describe the properties of magnets.
- Classify materials as magnetic or non-magnetic.

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- Explain how a magnetic object will become a magnet if we stroke it with a magnet.
- Identify that the Earth's magnetic south pole is near the geographic North Pole, which is why the north pole of a compass needle always points towards it.
- Define a magnetic field.
- Draw and interpret magnetic field lines around a magnet and between two magnets.
- Identify that the Earth's magnetic field is similar to the field produced by a large bar magnet.
- Describe how the Earth's magnetic field is produced by the movement of Earth's liquid outer core.
- Identify the components of an electromagnet.
- Explore how the number of turns in the coils, the electric current in the coil and the material of the core affect the strength of the electromagnet.
- Identify some applications of electromagnets.

Science 3

| Lessons | Learning objectives |
|--|---|
| Unit 1: Energy transfer | |
| Density of solids, liquids and gases Temperature and heat | Define density as the mass per unit volume of a substance. Calculate the density of an object by dividing the mass of the object by its volume. |
| 1.3 Explaining temperature and heat with the particle model | Identify the kg/m³ and the g/cm³ as density units. Predict whether a substance will float or sink when placed in water using the value of density. |
| 1.4 Heat and changes of | Define heat as the dissipation of energy from a hotter object or place to a cooler one. |
| state 1.5 Methods of heat transfer | Describe energy transformations using the law of conservation of energy. Differentiate temperature from heat. Explore how the change in temperature is affected by the mass of the object being heated. Relate temperature to the average kinetic energy of particles and thermal energy to the total kinetic energy of all the particles. Describe changes of state in terms of the arrangement, movement and energy of particles. Explore the factors that affect the rate of evaporation. |

| Uni | t 2: The Periodic Table | | | | |
|-------------------|---|--|--|--|--|
| 2.2 2.3 2.4 | The structure of the atom Electron arrangement and the Periodic Table Groups and periods in the Periodic Table Ionic bonds Covalent bonds | Recognise and describe the structure of a neutral atom. Define the atomic number and mass number of an atom. Calculate the number of subatomic particles in different atoms and isotopes. Relate the mass of an atom to the number of subatomic particles in its nucleus. Describe how the elements are arranged in the Periodic Table. Understand the link between the group number and the period number of an element and the arrangement of its electrons. Describe trends in properties of elements within groups in the Periodic Table. Describe how the properties of elements within periods in the Periodic Table. Relate the electron arrangement of elements within the same group to their properties. Explain how positive and negative ions are formed and how they form ionic bonds. Relate the structure of ionic substances to their properties. Describe how covalent bonds are formed using dot and cross diagrams. Explain what a molecule is. Relate the structure of covalent substances to their properties. Describe the structure of diamond and graphite as two different forms of carbon. Compare the physical properties of diamond and graphite in terms of their bonding and structure. | | | |
| | | Compare the properties of ionic to covalent compounds. | | | |
| Uni | it 3: Photosynthesis | | | | |
| 3.3 3.4 | The process of photosynthesis The structure of the leaf How photosynthesis can be affected Minerals for healthy growth | Describe how plants use light energy to produce carbohydrates through the process of photosynthesis. Write the word equation for photosynthesis. Explain why photosynthesis is important for life on Earth. Identify that photosynthesis happens in chloroplasts. Describe how leaves are adapted to carry out photosynthesis. Indicate the factors that affect the rate of photosynthesis. Experiment on how light intensity affects the rate of photosynthesis. Explain the importance of minerals for plant growth. | | | |
| Uni | t 4: Climate change | | | | |
| 4.2 | The recycling of carbon Changes in the amount of carbon dioxide in the atmosphere Effects of climate change | Describe how carbon is recycled by passing through organisms and depict this in flow diagrams. Explain the importance of carbon recycling for living organisms. Describe how fossil fuels are formed and identify combustion as the process that releases carbon dioxide into the atmosphere. Describe the key steps of the carbon cycle and depict them in diagrams of the carbon cycle. Identify human activities and natural processes that affect the carbon cycle. Describe how asteroid collisions with Earth cause climate change and mass extinction. Outline the impacts of climate change on the environment. | | | |

| Uni | t 5: Transport and excretio | n | | | | | | |
|--------------------------------|---|---|--|--|--|--|--|--|
| | Plants need water Transpiration Excretion in plants and | Describe how water is transported in plants through the roots, up the stem and finally to the leaves. Explain the importance of water in plants. | | | | | | |
| 5.4 | humans The urinary system | Describe the process of transpiration. Explain the effect of temperature, humidity, wind and light intensity on the rate of transpiration. | | | | | | |
| | | transpiration.Name excretory products of plants and the human body. | | | | | | |
| | | Identify the parts of plants and body organs responsible for excretion. | | | | | | |
| | | Differentiate solid wastes from excretory products. | | | | | | |
| | | Name the main parts of the urinary system and describe their function. | | | | | | |
| | | Describe ways to keep the urinary system healthy. | | | | | | |
| Unit 6: Reactions | | | | | | | | |
| | The reactivity series | Identify and explain how the reactivity series is constructed. | | | | | | |
| 6.3 | Displacement reactions Making salts | • Predict the position of a metal in the reactivity series based on its reactivity with water, oxygen or an acid. | | | | | | |
| 5.4 | Conservation during chemical reactions | Describe displacement reactions, and write their word equations. | | | | | | |
| | chemical reactions | Predict the outcome of a displacement reaction based on the reactivity series. | | | | | | |
| | | • Describe different methods of making salts and how to separate them from the other product(s) of the reaction. | | | | | | |
| | | Use symbol equations to describe chemical reactions. | | | | | | |
| | | Show how atoms are rearranged during chemical reactions. | | | | | | |
| | | • State the law of conservation of mass and use it to calculate the masses of the products and reactants of simple chemical reactions. | | | | | | |
| | | Describe how the law of conservation of energy is applied during chemical reactions. | | | | | | |
| Uni | 7: Human reproduction | | | | | | | |
| 7.1 The human reproduct system | | • Name the main components of the male and female reproductive systems and describe their functions. | | | | | | |
| | From fertilisation to birth Being healthy during | • Identify puberty as the stage in the life cycle of humans when the reproductive system of males and females becomes mature. | | | | | | |
| | pregnancy | Describe how the process of fertilisation occurs. | | | | | | |
| | | Outline the development of a foetus during pregnancy. | | | | | | |
| | | Indicate factors that affect the growth of the foetus. | | | | | | |
| Uni | t 8: Sound waves | | | | | | | |
| | Loudness of sound | Define amplitude of a sound wave and recognise it on a waveform. | | | | | | |
| | Pitch of sound Interference | Relate the loudness of a sound to the amplitude of the sound wave. | | | | | | |
| 0.3 | Interference | Define the frequency of a sound wave. | | | | | | |
| | | Compare the frequency of different sound waves by their waveforms. Relate the pitch of a sound to the frequency of the sound wave. | | | | | | |
| | | Describe the effects of wave interference and interpret waveforms of waves reinforcing or | | | | | | |
| | | cancelling each other. | | | | | | |
| Uni | t 9: Stars, the Moon and the Earth | | | | | | | |
| 9.1 | The formation of stars | • Identify that a nebula is a cloud of dust and gas, mainly hydrogen, in space. | | | | | | |
| 9.2 9.3 | The origin of the Moon Plate tectonics | Describe how stars are formed in nebulae. Describe different theories about how the Moon formed and state the evidence that | | | | | | |
| | | supports or contradicts them.Recognise that the giant impact hypothesis is the most prevalent theory for the formation | | | | | | |
| | | of the Moon. | | | | | | |
| | | State the evidence that supports there being tectonic plates and their movement on Earth | | | | | | |

| Unit 10: Electricity | | | | |
|--|---|--|--|--|
| 10.1 Electric current and | Recognise current as the flow of electrons and relate voltage to energy from the power | | | |
| voltage | supply. | | | |
| 10.2 Electrical resistance | • Identify the volt (V) as the unit of voltage and measure it with a voltmeter. | | | |
| 0.3 Series circuits | • Explore how adding cells affects current. | | | |
| IO.4 Parallel circuits | Define electrical resistance and explain its effect on charge flow in a circuit. | | | |
| 10.5 Everyday circuits | Calculate resistance by dividing voltage by current. | | | |
| | • Identify the ohm (Ω) as the unit of resistance. | | | |
| | | | | |
| | Apply Ohm's law to calculate current, voltage, and resistance. | | | |
| | • Measure current and voltage in series circuits and investigate how current is the same all around a series circuit, and how voltage across components in a series circuit adds up to the supply voltage. | | | |
| | • Explore how adding bulbs in series affects current, and give an explanation using resistance. | | | |
| | • Measure current and voltage in parallel circuits and investigate how voltage across the branches of a parallel circuit is the same and how current is shared between the branches. | | | |
| | • Explore how adding branches to a parallel circuit affects the current passing through the power supply and give an explanation using resistance. | | | |
| | Interpret, draw and construct circuit diagrams of everyday circuits. | | | |
| | Compare series and parallel circuits. | | | |
| Unit 11: Rates of reaction | | | | |
| I1.1 Determine the rate of | Define the term rate of reaction and describe ways of measuring the rate of reaction | | | |
| reaction | through the amount of reactant used or the amount of product formed. | | | |
| 1.2 Rates and concentration | • Draw and interpret graphs showing the production or consumption of a substance over | | | |
| 1.3 Rates and surface area | time. | | | |
| 1.4 Rates and temperature | • Explain why the rate of a reaction changes using the particle model. | | | |
| | • Explore how the concentration of a reactant affects the rate of reaction. | | | |
| | Use the particle model to explain why the rate of reaction increases when the | | | |
| | concentration of a reactant increases. | | | |
| | • Explore how the surface area of a reactant affects the rate of reaction. | | | |
| | Use the particle model to explain why the rate of reaction increases when the surface area | | | |
| | of a reactant increases. | | | |
| | | | | |
| | Explore how the temperature of a reactant affects the rate of reaction. | | | |
| | • Use the particle model to explain why the rate of reaction increases when the temperature of a reactant increases. | | | |
| Unit 12: Genetics | | | | |
| I2.1 Chromosomes | Describe the structure of chromosomes, and identify the genes as part of it. | | | |
| 2.2 Inherited and | Recognise that genes control a certain characteristic of the organism. | | | |
| environmental variation | State that a normal human cell contains 46 chromosomes. | | | |
| 12.3 Inheritance | Describe how we form a karyotype by arranging chromosomes in pairs. | | | |
| | | | | |
| | | | | |
| 12.4 Selective breeding | • Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. | | | |
| 12.4 Selective breeding12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. Describe the process of selective breeding. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. Describe the process of selective breeding. Outline the advantages of selective breeding. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explait how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. Describe the process of selective breeding. Outline the advantages of selective breeding. List the evidence that Darwin used to propose the theory of evolution by natural selection. Describe the process of natural selection, and explain how natural selection benefits | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. Describe the process of selective breeding. Outline the advantages of selective breeding. List the evidence that Darwin used to propose the theory of evolution by natural selection. Describe the process of natural selection, and explain how natural selection benefits organisms and leads to their evolution. | | | |
| 12.4 Selective breeding 12.5 Natural selection and | Define within-species variation and distinguish between variation caused by genes and variation caused by the environment. Recognise that mutations cause variation. Recognise that the X and Y chromosomes in humans are the sex chromosomes and explain how sex is determined. Use genetic diagrams to determine different characteristics and the sex of a human. Describe the process of selective breeding. Outline the advantages of selective breeding. List the evidence that Darwin used to propose the theory of evolution by natural selection. Describe the process of natural selection, and explain how natural selection benefits | | | |

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Questions about prior knowledge related to the course material of this unit.

Classification of living things

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Science around us

Some species show greater variation between individuals than other species. In some species, the individuals are very similar to each other. This can be caused by many different factors including human activity. The individuals of the European bison that live today are all offspring of 12 individuals that survived in the early 20th century. Because of extensive hunting, the European bison was almost extinct at that time. This means that the offspring which live today all have very similar characteristics to each other.

Another example of a species with individuals with very similar characteristics is the northern elephant seal. The northern elephant seal almost became extinct due to hunting and about 50 individuals survived in the early 19th century. All individuals of the northern elephant seal that live today are offspring of those 50 individuals and show a low variation between them. Low variation in a species can be dangerous for the survival of the species, since all individuals might not be able to adapt to environmental changes or diseases.

Revise your knowledge:

- What is adaptation?
- What is meant by the characteristics of an organism?
- What is a skeleton?
- What is the backbone?What is an exoskeleton?

Learning objectives:

- Identify living, once alive and never alive things based on the seven life processes.
- Classify organisms into plants and animals and these into smaller groups.
- Classify organisms into vertebrates and invertebrates and these into smaller groups.
- Define the term species.
- Identify variation within species.
- Use and create simple dichotomous keys to identify organisms.

A brief text that links the course material with everyday life.

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The key concepts students will learn in this unit.

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Science 1 • Student's Book • Sample pages

7.3 🔇 Classifying animals

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Vertebrates

Vertebrates belong to the animal kingdom, and they are all animals which have an inner skeleton with a backbone. There are different groups of vertebrates classified by their characteristics.



All the theoretical background the student needs is written in a concise text that contains all the necessary knowledge including terminology and fine points.

Questions that promote active learning and accompany the theory section.

In which group would you classify a crocodile?

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Invertebrates

Invertebrates are a group of animals of the animal kingdom. There are several groups of invertebrates with a variety of characteristics. However, all invertebrates do not have an inner skeleton or a backbone.

Molluscs: They are animals with soft bodies, but with strong muscles. Many of them have a shell which protects them. Sometimes this shell might be inside their body.

Jellyfish: They are animals that live in water. Their body is soft and jelly-like and they have long tentacles which have stinging cells. They use these tentacles to get their food in the water.

Annelids: They are worms, like the earthworm. Their bodies are soft and segmented.

Flatworms: They have soft flat bodies which are not segmented.

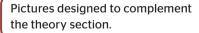
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Nematodes: They are worms that have soft and thin bodies, which are not segmented.

Echinoderms: They are animals that live in the water, like the starfish. Their body is hard on the outside and covered with spines.

Arthropods: They are animals with a segmented body and legs with joints. They also have a hard exoskeleton. There are many animals in this group. Two smaller groups of arthropods are arachnids and insects. Arachnids, like spiders, do not have antennae and they have eight legs in total. Insects, like bees, have antennae, they have six legs and they might also have wings.

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In which group

would you

classify an octopus?

Science 1 • Student's Book • Sample pages

Section for students to expand their acquired knowledge in different contexts.

7.3 S Classifying animals

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Dive into Science

What is a mus musculus?

Mus musculus is the scientific name of what we know as the common mouse. Scientists have been trying to classify organisms into groups for many years. Today they use a classification system which was invented by Carl Linnaeus, a Swedish scientist, who lived in the 18th century. He gave each organism a name with two Latin words, like the mus musculus. This naming system is still used today.



Activities

1. Answer the questions.

a. Which five groups of vertebrates do you know?

b. Which seven groups of invertebrates do you know?

2. Complete the table, as in the example.

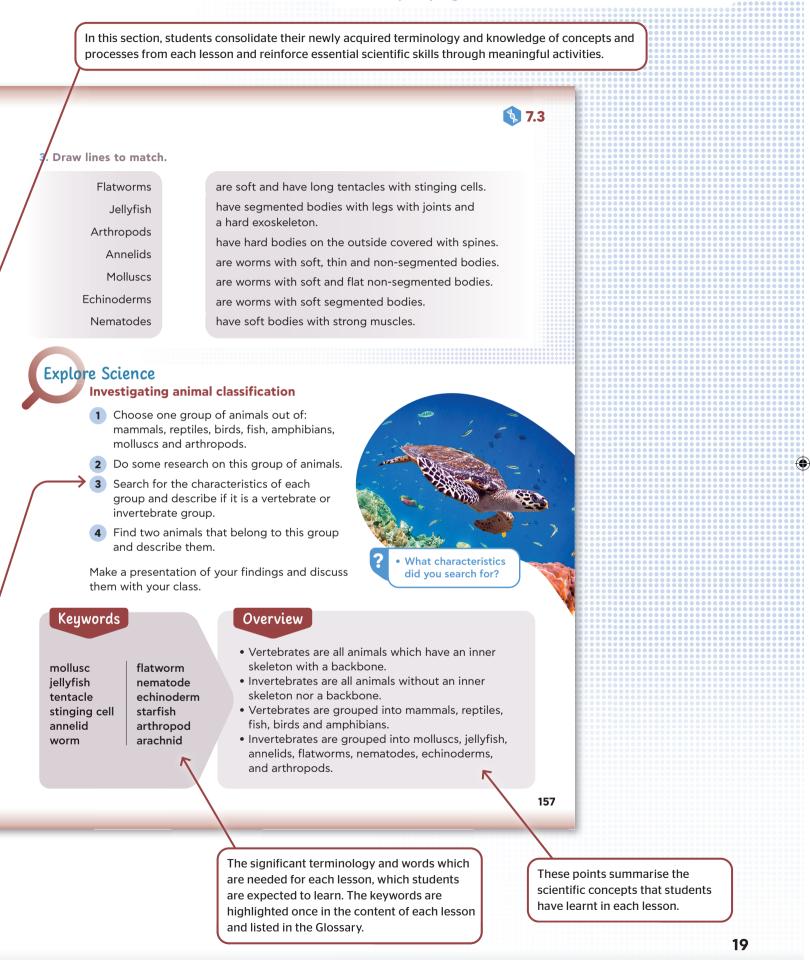
| Vertebrate groups | Characteristics | Reproduction | Breathing |
|-------------------|-----------------------|--------------|----------------|
| birds | feathers, beak, wings | lay eggs | lungs |
| | | give birth | |
| fish | | | |
| | scales, dry skin | | |
| | wet smooth skin | | skin and lungs |

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This section includes scientific enquiries, in which students have the opportunity to plan investigations, make predictions, observations and comparisons, use equipment to take measurements, record and display results, discuss and draw conclusions, construct models or do research about a topic. These investigations are usually carried out in groups or pairs. Teachers support students during these activities, but they do not transfer knowledge.

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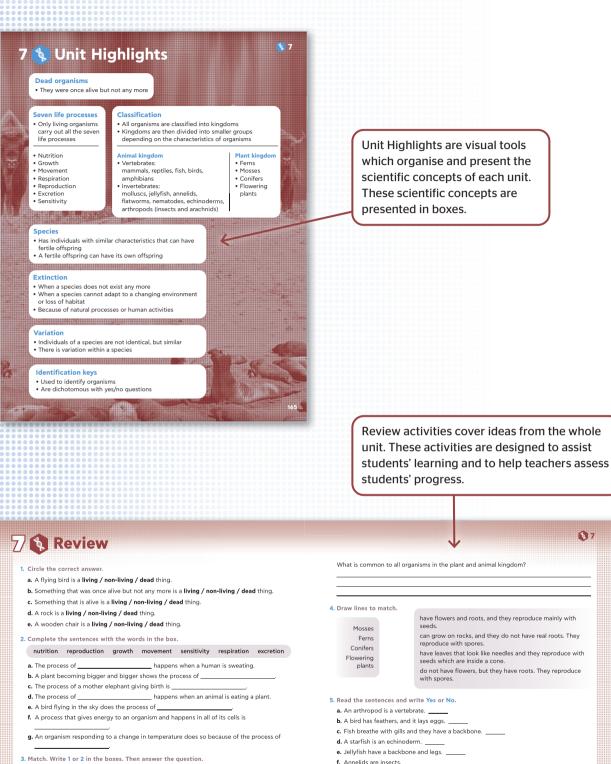
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ferns

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vertebrates (A)

conifers B

mosses

organisms that photosynthesise

organisms that move from place to place

 invertebrates
 F

 flowering plants
 G

organisms that eat plants or other animals

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1. animal kingdom

2. plant kingdom

g. Amphibians lay eggs in water.
h. Reptiles lay eggs in water.

6. Answer the questions

a. What is a species?

i. A spider is an arachnid not an insect. ____

b. What happens when a species goes extinct?

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Work like a scientist

Scientific method

In order to understand and explain the processes and things that happen around them, scientists base their thinking and research on the scientific method. Applying the scientific method as an investigation method ensures that scientists have a better chance of reaching reliable results to support or reject their hypotheses. The steps of the scientific method include:



3 S Form a hypothesis Research and gather information about the topic of your question and form a hypothesis that

4 B Test the hypothesis Make a prediction based on your hypothesis. Then plan an investigation to test your prediction to see if your hypothesis is supported by evidence or not. After that, carry out the investigation you planned and collect

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1 Observe The beginning of the scientific method is something you observe through your senses or even some background information you collect about a topic.

can be tested.

2 (?) Ask a question Ask a question based

on what you observe and you want to learn about. The question must be testable and should be answerable through scientific investigation. The question also should be exact and accurate and not answerable through ersonal opinion.

5 Description of the hypothesis is rejected, you must form another hypothesis and results and the hypothesis is rejected, you must form another hypothesis is rejected, you must form another hypothesis and repeat the process from step 3.

vour data.

6 B Share your results Report the results of your investigation and communicate them to others.

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A hypothesis can become a theory when a lot of evidence has been found to support it and it is repeatedly found to be accepted through different investigations from different groups. But even a theory does not always remain the same. A theory can evolve, adapt, be changed or be rejected and replaced by another theory.



The Work like a scientist section provides students with information about the steps of the scientific method. how to plan and conduct scientific investigations, how to take measurements and how to record and present their results. You can refer students to this section at the beginning of the year as well as during the year, as advised in the Teacher's Book or when you consider it would be to students' benefit.

All the laboratory equipment and special materials used throughout the book are presented with pictures to help students visualise them.

The Glossary contains all the keywords with comprehensible and age-appropriate definitions to ensure a gradual development of scientific terminology and mastery of communication skills. Additional vocabulary items included from lessons are marked with an asterisk (*).

Glossary

abbreviation*: a short way to write something abbreviation": a short way to write something absorb (for sound): to take in and stop sound instead of reflecting it or allowing it to pass through acid: a chemical substance that has a sourt sate and pH value between 0 and 7 acid rain: the rain that is more acidic than normal rain due to air pollution acidic: being a substance that contains acid;; being a solution of an acid adaptation: the fastures of a type of cell that are adaptation: the fastures of a type of cell that are

- alloy: a mixture of two or more different least one of which is a metal
- least one of which is a metal ammeter, a dwice that measures the size of an electric current **analogue:** a device that displays the measured value using a pointer that moves through a range of numbers **amelid:** a group of animals of worms with segmented bodies **arachnid:** a group of antimosods with eicht lens and

- arachnid: a group of arthropods with eight legs and without antennae. like a spider ngement: a specific way in which something is put in order
- put in order **arthropod**: invertebrate animal group, which includes insects, which have segmented bodies and legs with joints **ash**: the particles that are left after something has burnt
- burnt **asteroid belt**: a location between the orbits of Mars and Jupiter where we can find most of the asteroids in our solar system **atom**: the smallest particle of an element
- atom: the sinalest particle of an element atomic theory: the theory according to which the building blocks of matter are atoms **attract**: to pull something towards it **base**: a chemical substance that has a bitter taste, a soapy feel and a pH value between 7 and 14

- soapy teer alto a pri value between 7 and 14 basic: being a substance that contains bases; being a solution of a base batteyr: a source of electrical energy, made of two or more cells connected one after the other boiling: the change of state to a gas throughout the whole liquid 294

- boiling point: the temperature at which a liquid boils and changes into a gas bond (noun): a force that holds atoms together bond (verb): to join atoms chemically so they are held strongly together in a molecule
- brass: an alloy of copper and zind
- brass an alloy of copper and zinc break down to split a compound into simpler substances brittle: breaks easily into pieces bronze: an alloy of copper and tin buge: something that sticks out from a surface and is round in shape burette: special laboratory equipment used for adding measured volumes of a liquid
- adding measured volumes of a liquid carbon dioxide (CQ) is a gas that is in the air; a compound of carbon and oxygen atoms carnivers: an animal that eats only animals cell: the smallest building unit of living organisms; cell: the smallest building unit of living organisms; vincit component that stores chemical energy which can be transferred as electrical energy round the circuit
- cell membrane: a thin layer on the outside of the cell that controls the flow of substances into and out of the cell
- cell wall: a strong part of certain cells found on their outside (e.g. in plant cells) cellulose: a substance from which cell walls of plant cells are made

- cells are made cells are made cementation: the process by which dissolved materials like sediments stick together forming sedimentary rocks chemical energy: the energy that is stored in fossil fuels, food, batteries, etc. chemical formula (pl. chemical formulae): a way to represent the molecule of a substance, using chemical symbols and the number of the atoms the molecule consists of written in subscript -band-al procestry: a corcept what can only be
- chemical property: a property that can only be observed and measured through chemical
- changes changes chemical reaction: an event in which one or more substances turn into one or more new substances chemical symbols a unique symbol, formed from capital and lowercase letters of the latin alphabet, used to represent each element cherophylls substance which is important for photosymbols and which makes chloroplasts cheroplasts a part of some plant cells in which botosymbols hopens and which is oreen
- chloroplast: a part of some plant ceils in which photosynthesis happens and which is green because it contains chlorophyll cilium (pl. cilia): structures that are part of the cytoplasm of some cells and which look like hairs

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Science 1 • Workbook • Sample pages

7.3 Classifying animals

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 Look at the pictures and tick (v) the vertebrates. Then write the name of the vertebrate or invertebrate group the animal in the picture belongs to, as in the example.



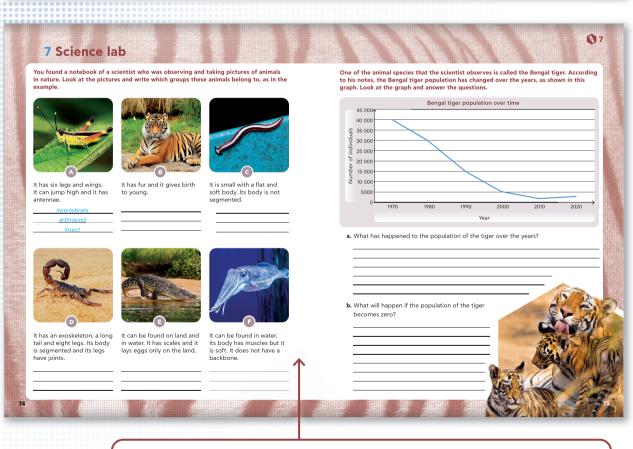
6 7.3

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5. A spider is not an insect. Why?

2. Complete the sentences with the words in the box



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At the end of each unit the Workbook contains a Science lab activity which the teacher can assign as homework. This activity helps students be more creative and use all the knowledge they have gained from the unit in a more complex and combined activity.

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Unit map 🔹 ႗

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Classification of living things

Scientific information

Our planet is about 5 billion years old and life on Earth has been around since about 35 billion years ago. It is estimated that loday there are about 8 million species on our planet that can be found from the deepest oceans to the highest mountains. There is o much diversity between living organisms that scientists have been classifying organisms for was is noted to sciul them and they dish were measing on the science of the scie ars in order to study them and they still haven't classified em all. Classification happens through the comparison of the organism to another, however it is not as simple as it ight sound. There is a need to have a universal classification stem to recognise organisms and scientists have had sagreements as to how to create such a system. cation diagreements as to how to create such a system. The system of classification agreed upon and used doty its based on the work of Carolus Linnaeus who lived in the 18th century. In this system, all kinding organisms are classified into different subdivisions according to their characteristics. The largest division is into kingdoms. Linnaeus only described bwo kingdoms, the plant and the animal kingdom, however today with the development of technologies little the microscope there are more kingdoms there are for a long thm. It more one how many liquidoms there are for a long thm. It more animals, plants, turing, innorer (prokaryote), and potsta. animals, plants, fung, incores (prokaryotek), and protista. After the division to kingdoms, organism are further divided into privium, class, corter, family, genus and finality into the smallest taxonomic group of species. According to this classification system, each organism has a unique Latin name consisting of two works. For example, fundamina era mand *Homo* salpiers, Homo is the genus and saptens is the species. In the example of the common mouse, it is called Mas musculus and its classification is the following kingdom: annia, phytum-chotak, class: mammals, order: rodents, family, Murdae, genus, Mas, species: musculus. The classification is bade on the characteristics of each species. These are annaler subgroups after the classification into specing and the the species of by humans such as pels like cats and dogs. However, breeds are not officially

recognised classification divisions, instead they are practical terms. Defining a species is not an easy task. In fact, It has been a very tard task for selectists to define species. The definition of a species is a group of animals with similar characteristics which matches there are many saves with this definition. First, and the sent many saves with this definition, first, and the sent many saves with this definition, first, and the sent of the sent end with the set of the past are offspring. Second, species evolve. The species of the past are offspring. Second, species evolve. The species of the past and the same tacky loculution is key in understanding how species change and evolve with time. Today, some of the NoI of each organism is be sequencing, cleantists use the NoI each organism. The organisms. Buy can taked its sequence and compare it to other organisms. Sur-stragentism and are biodiver with the comparisms has shared integrations and a the local scheduler safety has a single and the same through the years. The classification of hioms staged the same through the years. The classification of hioms according to their characteristics in order to study them in detail. detail

Language focus

There are plenty of new terms in this unit. Spend the time needed to study them in detail and ensure you illustrate their use orally and in writing during the following lessons, Give Ss motivation and opportunities to use these terms themselves.

nnelid: a group of animals of worms with segm annelid: a group of animals of worms with segmented bodie arachnid: a group of arthropods with eight legs and without antennae, like a spider arthropod: invertebrate animal group, which includes insect which have segmented bodies and legs with joints

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The first pages of each unit contain all the scientific background information that a teacher may require in a lesson.

The language focus section contains all the keywords with definitions. Additional vocabulary items included in lessons are marked with an asterisk (*). At the same time, we inform the teacher about possible problems that may arise when using some terms in the Be aware section.

7 🗞 Unit map

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- spikes end covered with body cover a reproduce with spress and cover with similar discretions the spress of removing wate products from their environment discretions the station in which there are not any individual spress are rootable with spress and cover with similar testers in their environment discretions the station in which there are not any individual spress are rootable with spress and cover and base start share were with similar from agroup of plants which reproduces with spores and base star share were with spress and base star share were with spress and base as the satisfies which against the spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable with there are not any individual spresses are rootable to reproduce and reservements are children and has a star share were maint that lives in the reservements are children and has a star share were maintent and the spresses are rootable to reproduce and reservements are children and has a star share were maintent and the spresses are rootable to reproduce and reservements are children and has a star share were maintent and the spresses are rootable to reproduce and reservements are children and has a star share were maintent and the spresses are rootable to reproduce and reservements are children and the spresses are rootable to reproduce and reservements are children and the spresses are rootable to reproduce and reservements are children and the spresses are rootable to reproduce and reservements are children and the spresses are rootable to reproduce and reservements are children and the s
- not nave inverse fertile: being able to reproduce and create new organisms flatworm: a group of worms with flat non-segmented bodies growth: the process of changing to be larger in size, height and weight
- and weight lidentifications the same in every detail identifications the process by which something is recognised identification weight a war to identify living organisms using questions at each step about their characteristics individual: one single organism intertifient on being able to reproduce and create new organism
- ertebrate: an animal that does not have an inner skeleton with a backbone
- lyfish: Invertebrate group of animals that live in the water and have a soft and jelly-like body with tentacles and stinging cells
- gdom: the largest group into which living organisms are classified
- ic: invertebrate animal group with a soft body with ong muscles
- a group of plants that do not have real roots or flowers nd reproduce with spores

- sport: a reproductive body of some plants like mosses starlish: an investbraite echnologen animal that lives in the sea and has a size shape usually with five arms stinging cells a type of specialised cell found in some animals that contains chemical which can harm other organisms when in contains chemical which can harm other organisms that contains chemical which can be some in iself with testacter along and its free might seem like arms or antennae wination offerencis in the characteristics of organisms that with otherwise look very similar with otherwise look very similar
- vertebrate: an animal that has an inner skeleton with a backbone
- rm: a type of invertebrate animal without arms or legs that

Be aware

Betwarre
 The word registation' can have different meanings. In this um, "registration' can bave different meanings. In the special system of the special of the special cod which happens in every cold an organism. However, in everyday life this word can also mean the process of breathing.
 The word 'kingdom' can have different meanings. In biology 'kingdom' is a classification term of a large group of organisms with certain similar characteristics, like the plant kingdom and the animal kingdom. However, in everyday life the word is also used to describe a territory governed by a king or queen.

Unit map 🐧 7

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7.1 The seven life processes this lesson, Ss will recognise the difference between organisms that are alive, organisms that were once alive and objects that are never alive. Then they will extend their knowledge of the seven life processes that define a living organism.

Identify living, once alive and never alive things based on the seven life proce ntific Enquiry Skills stigate phenomena, objects, materials and living things in order to sort and classify then

For the presentation of the keywords, see the guidelines in the TB map. • nutrition > growth > movement > respiration > reproduction > excretion > sensitivity Cross Curriculum Links (CCL)

This lesson can be linked with lesson 3.1, as Ss already know that living organisms are made of cells which are the smaller units of life.

Materials and Resources

> Its Allve or not, W5 b More practice > Explore Science. A3 white card (1 piece per group), coloured markers > More parcicle for higher-performing 5s: plastic cup (1 per 5), linewater solution (50 m/ per 5), test tube (1 per 5), straw (1 per 5)

Com on Student Preconceptions

Ss may think that a characteristic of living things is breathing air.
 Some Ss may confuse respiration with breathing.
 Some Ss may not understand that plants also carry out all seven life pro-

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The next pages contain an informative overview of the lessons of each unit. The learning objectives, scientific enquiry skills, keywords, Cross curriculum links, materials and resources needed as well as a list of students possible difficulties and/or preconceptions are provided in well-organised tables that enable the teacher to monitor the progression of knowledge and observe the structure of the scientific content.

Science 1 • Teacher's Book • Sample pages



Science around us

around us section. Have a S read aloud each paragraph of the Science around us section. Be ready to support Ss with the pronunciation of the scientific words. pronunciation of the science was Make sure a different group of Ss read the text aloud each time so the everyone has the chance to read throughout the school year.

Revise your knowledge:
 Ask Ss What is adaptation? (Accept all reasonable answers. Suggested answer: Adaptation is the changing of characteristics and/or the way an organism responds to things so that the organism can survive in an

environment.). What is meand by the characteristics of an organism? (Accept and irresonable anxies. Suggested answer: A characteristic is something thar makes an organism different from anothe.). What is a selector? (Assident) as all those insteads (Assident) as all the distance of the instead (Assident) as all the distance of the instead (Assident) as all the distance of the instead (Assident) (Asside
 backbone (The backbone is the line of bones in an animals or human's back that are connected together J, What A that are connected together J, What A and order and that Covers, support and support and that the support and support and that the support and support and that the support will be about.

s read aloud the learning res that will be covered in this

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After the aggregate information about the unit, the presentation of the unit begins with the cover page. Here we can find the answers to the questions shown in the Revise your knowledge section.

7.3 🔇 Classifying animals



Invertebrates

Draw Ss' attention to the theory section Invertebrates.
 Explain to Ss that invertebrates belong to the animal kingdom and that they do not have an inner skeleton or

backbone. Point out to Ss that there are seven groups of invertebrates described in this lesson. Focus Ss' attention on picture F. Explain to Ss that in picture F, we can see an example of a molifier.

Explain to Ss that in picture is, we can see an example of a moliaic.
 Explain to Ss that a moliusc has a soft body with strong muscles and Gometimes) a protective shell.
 Focus S's attention on picture G.
 Explain to Ss that in picture G, we can see an example of a jellyfah.
 Explain to St that in picture G, we can see an example of a jellyfah.
 Explain to St that in picture S, we can see an example of a picture St the interpicture S.

Focus Ss' attention on picture H. Explain to Ss that in picture H, we can see an example of an

 Explain to Ss that annelids are worms with soft segmented Focus Ss' attention on picture I.

Explain to Ss that in picture I, we can see an exa natworm. Explain to 5s that flatworms have soft flat bodies which are not segmented. Focus Ss' attention on picture J. Explain to 5s that in picture J, we can see an example of a nematode. • Explain to Ss that nematodes are worms with soft thin bodies that are not segmented. • Focus Ss' attention on picture K. • Explain to Ss that in picture K, we can see an example of an orbitenderm echinoderm. Explain to Ss that echinoderms live in water and their body is hard on the outside and covered with spines. Focus Ss' attention on picture L. Explain to Ss that in picture L, we can see an example of an Logian to Sa that in picture L, we can see an example of an arthropod.
 Explain to Sa that arthropods are animals with segmented bodies and joints in their legs.
 Point out Sa that arthropods an intrifuent divided into advantation of the second Invertebrates. Focus Ss' attention on picture M. Explain to Ss that in picture M, we can see an octo

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• Accus sa attention on picture M. Explaint to St that in picture M. Accus see an octopus. Ask St in which group work group classify an octopus? Ask St in which group work group classify an octopus? In the second sec ng cells.). Ss some time to think about their answers. rrage Ss to express their opinions, and initiate a short

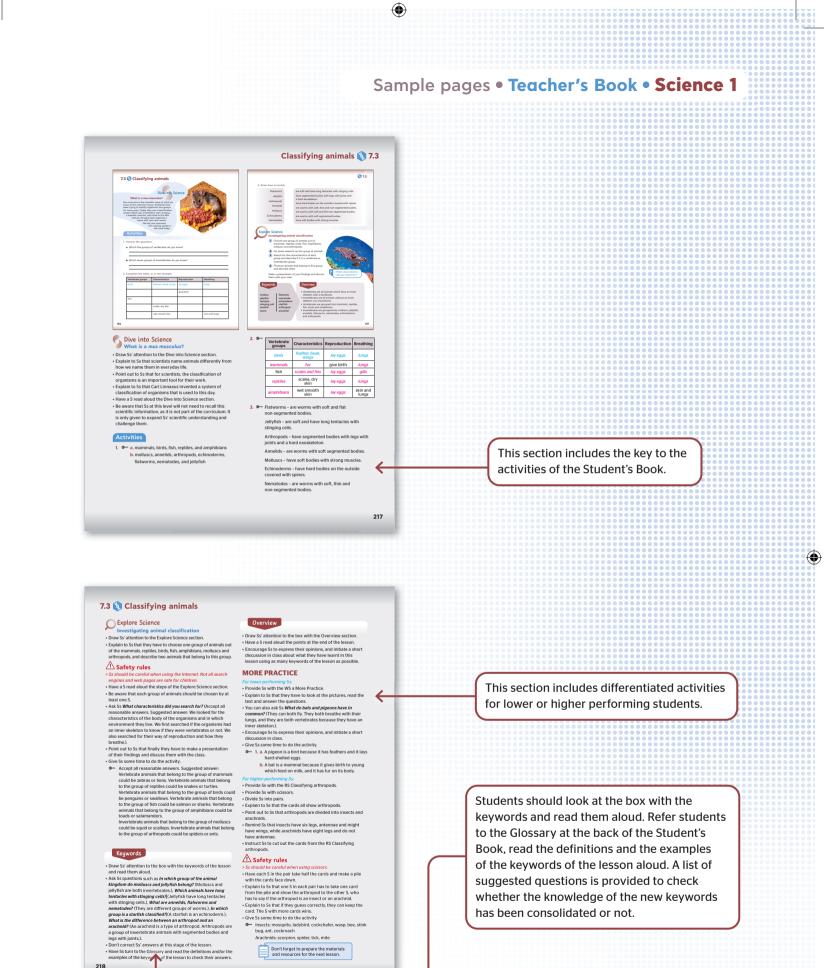
This section refers to the corresponding theory sections. During this phase, the teacher introduces the topic, while students begin to engage with the topic through questions, discussions, etc. Afterwards, questions arise in students' minds and prior knowledge (including preconceptions) is accessed. The identification of students' current conceptions and preconceptions is very important for conceptual development to take place. Students are encouraged to explain concepts in their own words and to listen critically to each other's explanations. Meanwhile, teachers can introduce specific scientific vocabulary and help students to use this vocabulary to explain the results and justify their explanations. Students should feel free to express themselves, and teachers should accept all thoughts, views and opinions expressed by students.

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Don't forget to prepare the materials and resources for the next lesson.

Science 1 • Teacher's Digital Material • Sample pages

Worksheets and Resources are provided as additional supportive material that reinforce and expand newly acquired knowledge.

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Worksheet a More practice 7.3 Classifying animals Name: Date: 1. Look at the pictures, read the text and answer the questions. Worksheets follow a well-defined and structured template for a comprehensive learning experience A pigeon has feathers and lays eggs with a hard shell. A pigeon can fly from place to place using its wings. A bat can also fly from place to place, but it has fur on its body, and it gives birth to young, which are fed with milk. € and they also support experimentation and investigation activities. a. In which group of vertebrates would you classify a pigeon? Explain your answer. b. In which group of vertebrates would you classify a bat? Explain your answer 52 Resource Sheet Food chain Resource Sheets serve as an additional reference point with pictures and extra information useful for more practice sections. 26

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Sample pages • Student's Book • Science 2



Facts about asteroids

Our solar system consists of millions of asteroids while more than 1 000 000 of them are known. Asteroids are lumps made of rock with an irregular shape. They are generally are generally in colour and most of them have <u>craters</u> on their surface resulting from collisions with other objects.

objects. Asteroids vary in size. Their diameters range from a few metres for the smallest, to a few hundred kilometres for the largest. The largest asteroid found so far is the asteroid Vesta which is close to 525 kilometres in diameter. Asteroids along with comets and meteoroids are some of the smallest objects that make up our solar system.

Example

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b. ____

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How many times wider is the Moon than Vesta, if the Moon's diameter is 3475 km? Round of your calculations to 2 decimal places. if the Mo

To find how many times wider the Moon is than Vesta, we should divide the diameter of the Moon by that of Vesta. So, <u>3475</u> = 6.62. Theref ore, the Moon is approximately 6.62

imes larger than Vesta

Asteroids were formed in the early stages of the formation of our solar system. These small pieces of rock are the leftovers of the nebula from which our solar system was born 4.6 billion years ago, which did not manage to grow large enough to become planets. Some asteroids were part of larger objects, like planets, moons, or bigger asteroids, that broke off after colliding with another object.

Since asteroids are small objects, compared to Since asteroids are small objects, compared to planets, their gravity is small and therefore the materials, i.e. the rocks, from which they are made are not strongly held together. This results in their shape sometimes changing due to the tidal forces exerted on them by larger solar system objects. Also, only a few large asteroids have their own moons, e.g. Didymos with its moon Dimorphos.



Some asteroids may be made of different pieces of rock, while others that are small and spin quickly are made of only one piece of rock. This can be explained because their low gravity would not hold different pieces of rock together

Asteroids orbit the Sun. The majority of asteroids known to us are located in the **asteroid bell** between the orbits of Mars and Jupiter while other share the orbit of a larger planet. Jupiter has one of the largest and most important populations of such asteroids known as Jupiter trojans. However, Jupiter is not the only planet with this characteristic. Even Earth has two asteroids with which it shares its orbit, with the first discovered in 2011 and the second in 2021. In addition, there are also some asteroids whose orbits pass close to (Near-Earth Asteroids) or even cross Earth's orbit (Earth-crossers).

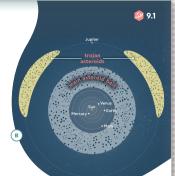
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Asteroid impact on Earth

In some cases, the orbit of an asteroid can change. When an asteroid passes dose by a planet or a massive object, the force of gravity exerted by the planet on it can pull the asteroid out of its orbit. For example, it is possible for the Earth to exert such a large force of gravity on an asteroid that it pulls it towards it. If this happens, an asteroid impact may corur on Earth's surface. Another way for an asteroid impact to happen is with an asteroid whose orbit crosses the Earth's orbit.

One asteroid impact with Earth, known as the Chicxulub impact, is responsible for the extinction of the dinosaurs and 75% of species 66 million years ago. On the other hand, another asteroid impact with Earth is also believed to have been the cause of the star of life on Earth. Asteroids hit the Earth very

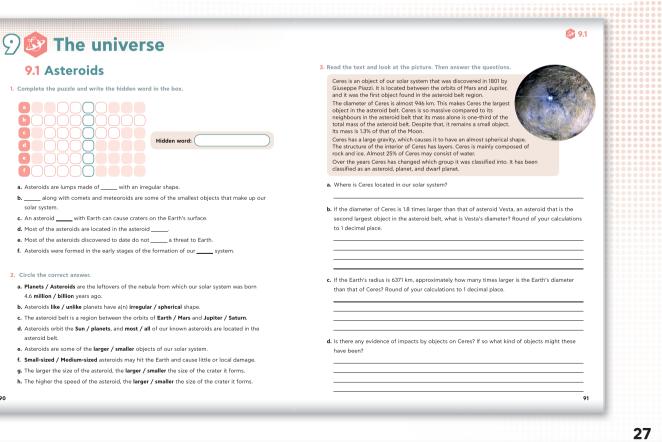
frequently. Whethe an asteroid coming towards Earth po



a threat depends on its size. Most asteroids a threat depends on its size. Most asteroids that come towards Earth are small enough to burn up in the Earth's atmosphere, and thus do not reach the surface, so they are not considered dangerous. However, medium-sized asteroids may hit the Earth and cause little or local damage to the area while large-sized asteroids may cause extinction.

Sample pages • Workbook • Science 2

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Science 3 • Student's Book • Sample pages

8.3 🛞 Interference

Wave interference

When a drop of water falls in water, a wave spreads out on the surface of the water. When a series of drops fall into water, waves spread out on the surface of the water forming circles around the point the drops fall. We can use this as an analogy of how sound waves travel in a medium, since sound waves cannot be seen.



be seen. Two or more waves from different sources can meet while travelling through the same medium. This causes the particles of the medium to vibrate in a way that results from the effect of both waves. This is a phenomenon known as wave interference. The amplitudes of the waves that interfere add together. Since sound waves are invisible, the effect of interference can more easily be seen by analogy with water waves. There will be some points that vibrate with a higher amplitude, and others that stay still.

Although interference happens between any waves of the same type, e.g. between two sound waves, we will only examine waves of the same frequency and amplitude that interfere with each other.

Types of interference

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Constructive interference If two waves, coming from two different sources, meet while travelling through a medium and their peaks line up with peaks and troughs line up with troughs, they reinforce each other. This type of interference is called **constructive** interference. In the case of constructive interference between two sound waves this means their compressions line up together and their rarefactions line up together. Constructive interference results in a wave that has the same frequency as the waves that interfere and an amplitude equal to the sum of the two waves' amplitudes.

How is the loudness of the sound affected when sound waves reinforce each other?



Destructive interference Waves, coming from two different sources, can also meet in such a way that their peaks line up with troughs. This type of interference is called destructive interference is called the amplitude to decrease. In the case of distructive interference between two sound waves, the compressions of one sound waves in the campart of the other sound wave. Destructive interference results in a wave that has the same frequency as the waves that interfere and an amplitude equal to the difference between the two waves' amplitudes. If the waves that interfere have the same amplitude, they gancel each other out source is called and the amplitude of the wave produced is zero.

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Uses of interference

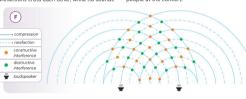
In concerts, at least two loudspeakers produce sound waves that travel through air in front of the loudspeakers, spreading out into the area. The sound waves interfare in such a way that there will be positions of louder sounds and positions of no sounds. The louder sounds are heard where two waves reinforce each other, i.e. where two compressions or two rarefactions cross each other, while no sounds are heard where two waves cancel out, i.e. where compressions of one sound wave cross rarefactions of the other sound wave. Even though the sound waves coming directly from the two loudspeakers are cancelling each other out in some positions, you will still hear some sound in these positions, because sound waves also reflect off the walls, objects and even the people at the concert.

How is the loudness of the sound affected during destructive interference?

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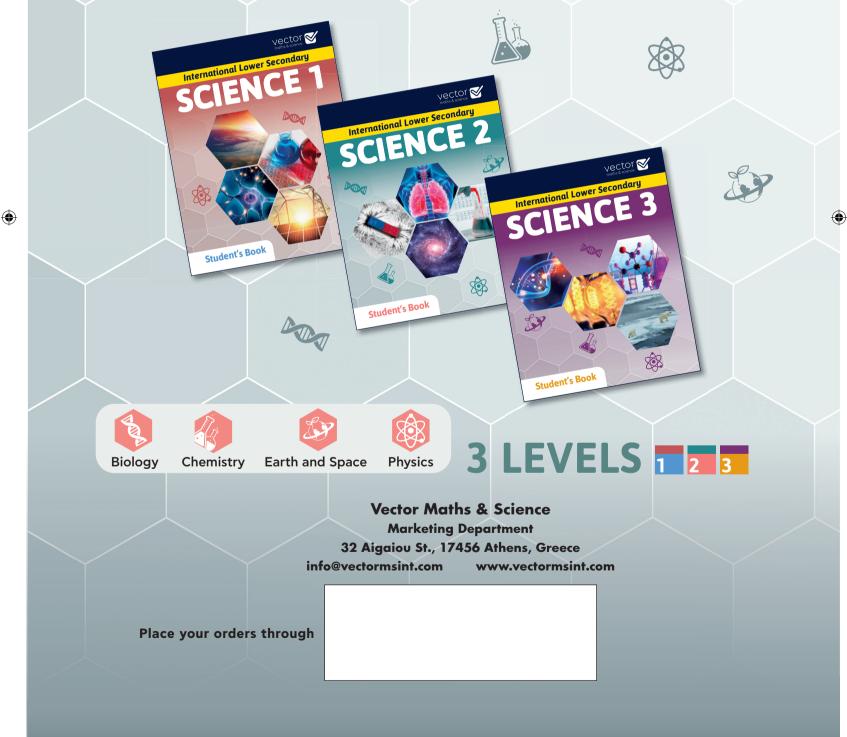
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Science 3 • Workbook • Sample pages

🚯 8.3 8.3 Interference 1. Circle the correct phrase to complete the sentences a. You hear a note played on two guitars at the same time louder than they are heard separately because of 1. constructive interference 2. destructive interference b. The frequency of two sound waves that reinforce each other 1. stays the same 2. increases c. Sound waves can cancel each other out if _____ 1. they have double the frequency 2. they have the same amplitude d. At a concert, there are positions where you can hear the sound louder, and positions 1. you cannot hear any sound 2. you can hear the sound softer 3. Look at the pictures and answer the questions. e. Theatres are designed so that _ 1. destructive interference is prevented 2. constructive interference is prevented a. Why might some people hear louder sounds than others at a concert? f. Noise-cancelling headphones work by making sound waves 1. reinforce each other 2. cancel out 2. Look at the waveforms of different waves that interfere and draw the result of each wave interference. b. Why do theatres have soft seats, curtains and carpets? 88 89



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